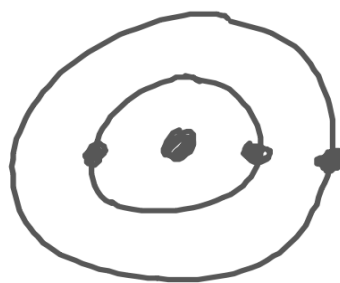


${}^6_3\text{Li}$

$\frac{3}{3}$ p^+
 $\frac{3}{3}$ n^0
 $\frac{3}{3}$ e^-
 1 ne^-



$\text{Li} \cdot$

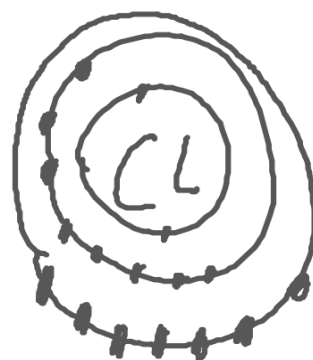
~~Chlorine~~ Chlorine

17 p+

16 n^o

17 e⁻ 3s

2 Ne-17 Cl



1

2 electrons



S-Level

Bohr said $2e^-$ on 1st level

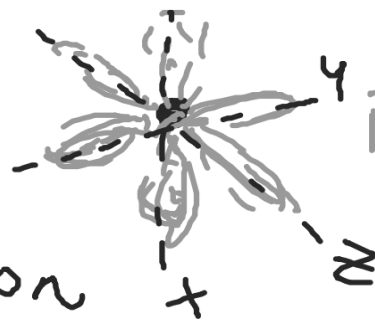
2

2 electrons



S-level

6 electrons



P-level

Bohr said $8e^-$ on 2nd level

[3] 2 electrons S-Level

6 electrons P-level

10 electrons d-level

Bohr said $18e^-$
On 3rd level



1s

2s

2p

3s

3p

4s

3d

4p

5s

4d

$1s^2 2s^1$

Li 

Cl $1s^2 2s^2 2p^6 3s^2 3p^5$

N
7e- $1s^2 2s^2 2p^3$